

# Defining boundaries: which *N*-nitroso compounds might not belong in the cohort of concern?

Dr Robert Foster, Dr David Ponting.

Granary Wharf House, 2 Canal Wharf, Leeds, LS11 5PS

## ■ Cohort of concern: a vague definition

*N*-Nitroso compounds belong to the cohort of concern due to some members of the chemical class having significant carcinogenic potential below the general 1.5 µg/day threshold of toxicological concern (TTC) limit. Structurally-defined by name only, *N*-nitroso compounds are broadly represented by N=N=O; however, around half the *N*-nitroso compounds analysed have lower carcinogenic potency than the TTC [1]. In fact, high potency is attributed to those capable of undergoing facile  $\alpha$ -hydroxylation & subsequent formation of a diazonium ion in proximity to react with DNA. An understanding of the structural features which mitigate activity has been curated [2, 3] & used to suggest refinement of the structural definition of the cohort of concern to allow *N*-nitroso compounds with low carcinogenic risk to be assessed in the same manner as other impurities under ICH M7 without increasing the risk to human health [4].

## ■ $\alpha$ -Hydroxylation: the potent mechanism

The carcinogenic activity of *N*-nitroso compounds is typically attributed to metabolic formation of an alkyldiazonium ion *via*  $\alpha$ -hydroxylation (Fig. 1). Although it is possible for carcinogenicity to occur *via* other mechanisms (e.g., nitrosoureas may react directly with DNA, whereas *N*-nitrosodiarylamines may undergo transnitrosation to *N*-(nitrosoaryl)aniline), significant carcinogenic potency is linked to the ability to undergo facile  $\alpha$ -hydroxylation.

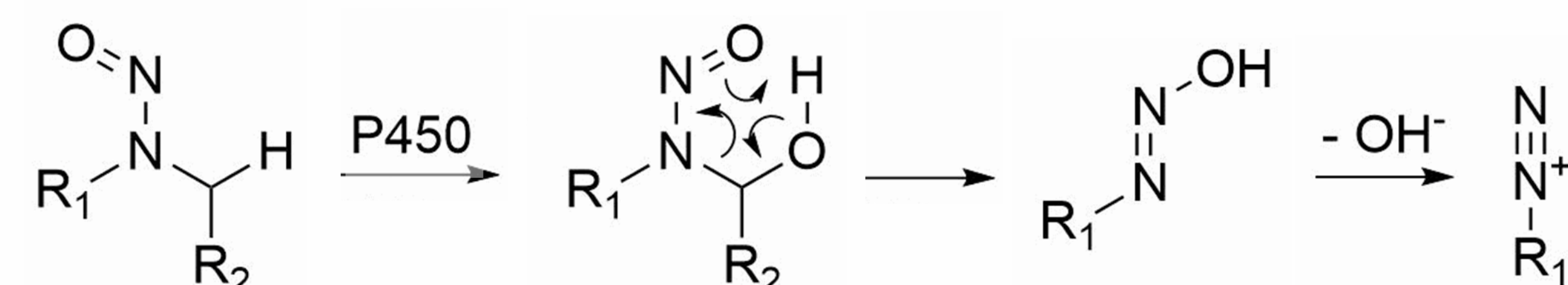
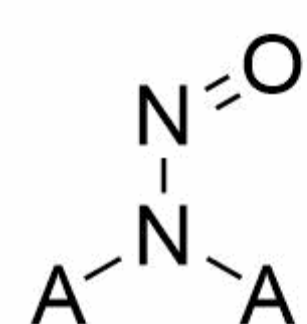


Fig. 1. Alkyldiazonium formation *via* *N*-nitrosamine  $\alpha$ -hydroxylation.

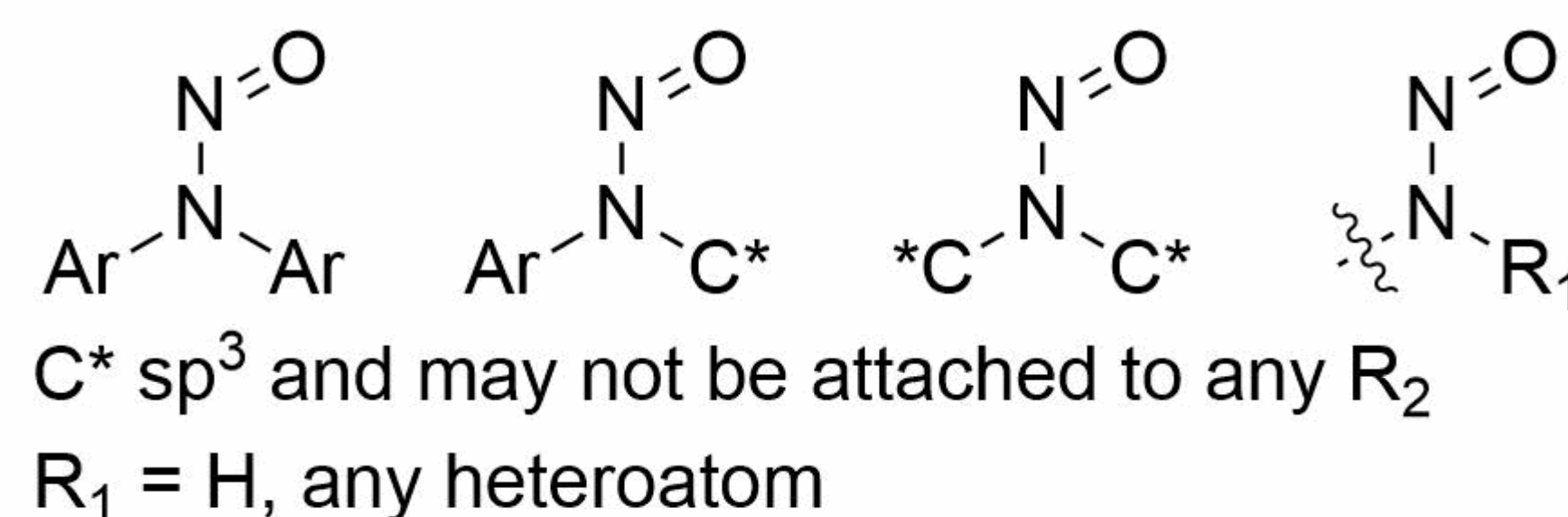
## ■ Cohort of concern: redefined

Presence of an  $\alpha$ -hydrogen is critical for significant carcinogenic potency. Therefore, it may be reasonable to redefine the cohort of concern to include only dialkyl *N*-nitrosamines bearing an  $\alpha$ -hydrogen, or a hydroxy or an alkoxy substituent (as the metabolic products) (Fig. 2). Nitrosoureas & analogues may be conservatively included due to comparable potency; however, a negative result from the Ames test should be considered acceptable.

### Current cohort



### Subclasses to remove from cohort



### Proposed cohort

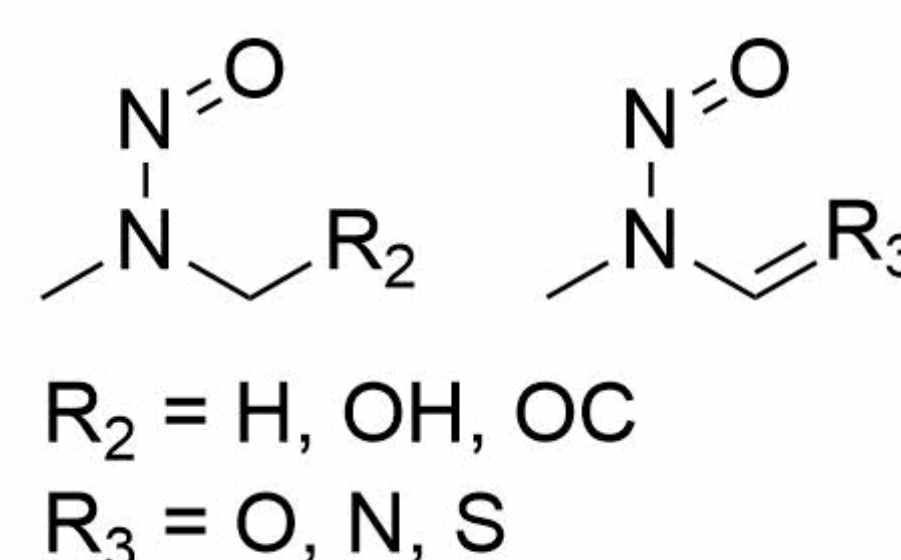
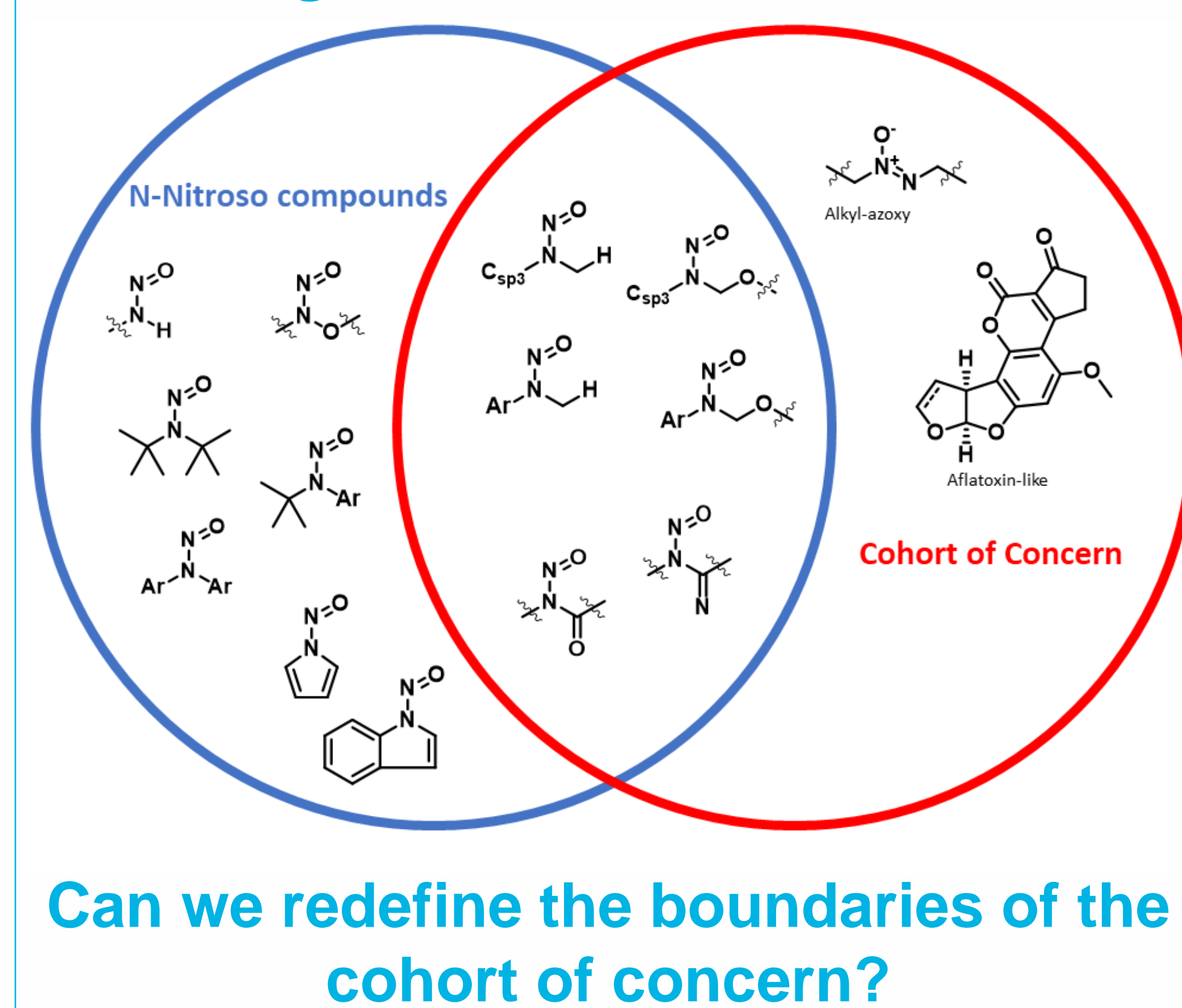


Fig. 2. Current scope of the cohort of concern, subclasses to remove from the cohort of concern & the proposed scope of the cohort of concern.

## ■ *N*-Nitroso compound subclasses: cohort or not?

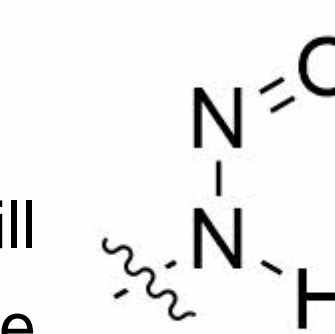
The rate at which metabolic activation occurs (if it is even possible) & the stability of the resulting alkyldiazonium ion is not consistent between subclasses of *N*-nitroso compounds [4]. Fig. 3 briefly highlights those subclasses which may be considered outside the scope of the cohort of concern due to their lack of significant carcinogenicity.

### Which *N*-nitroso compounds might not belong in the cohort of concern?



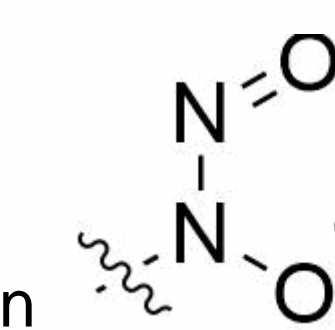
### 1. Nitrosated primary amines

- May form diazonium ion *via*  $\alpha$ -hydroxylation; however, it will be inherently unstable *in vivo* & is expected to decompose before reaction with DNA can occur.



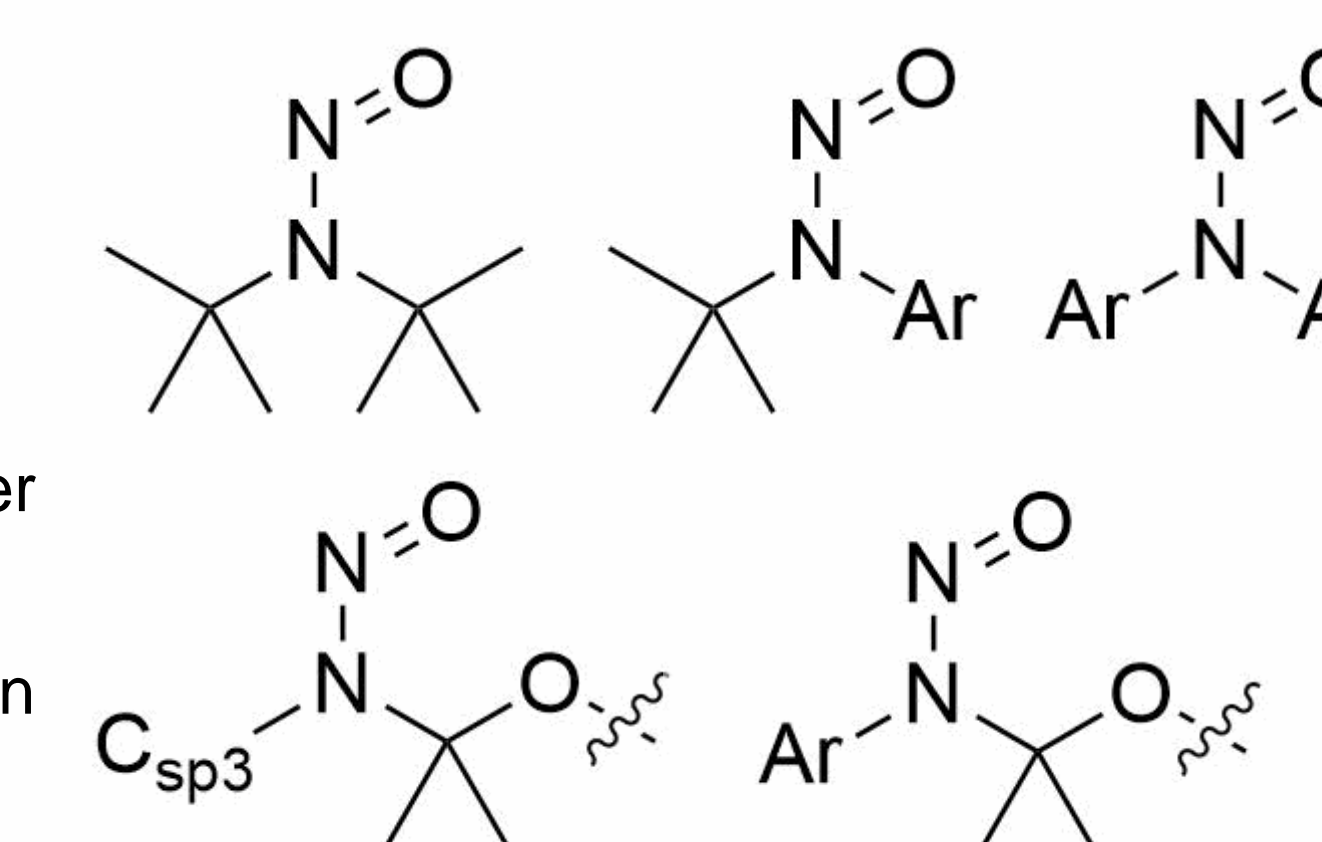
### 2. Nitrosoalkoxylamines

- May undergo  $\alpha$ -hydroxylation but are incapable of forming an alkyldiazonium ion & are observed to be weakly carcinogenic or non-carcinogenic [3].



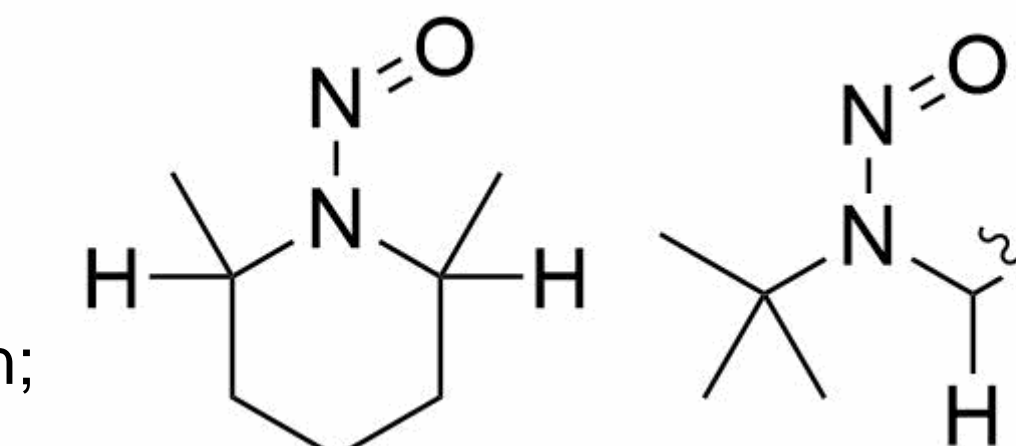
### 3. Nitrosamines without $\alpha$ -hydrogen

- Cannot undergo  $\alpha$ -hydroxylation to form a diazonium ion.
- Aliphatic substituents: negative in carcinogenicity assays.
- Aromatic substituents: weakly carcinogenic *via* other mechanisms (e.g., transnitrosation to *N*-(nitrosoaryl)anilines).
- $\alpha$ -Hydroxy/ $\alpha$ -alkoxy: metabolic intermediate in  $\alpha$ -hydroxylation mechanism & observed to be carcinogenic.



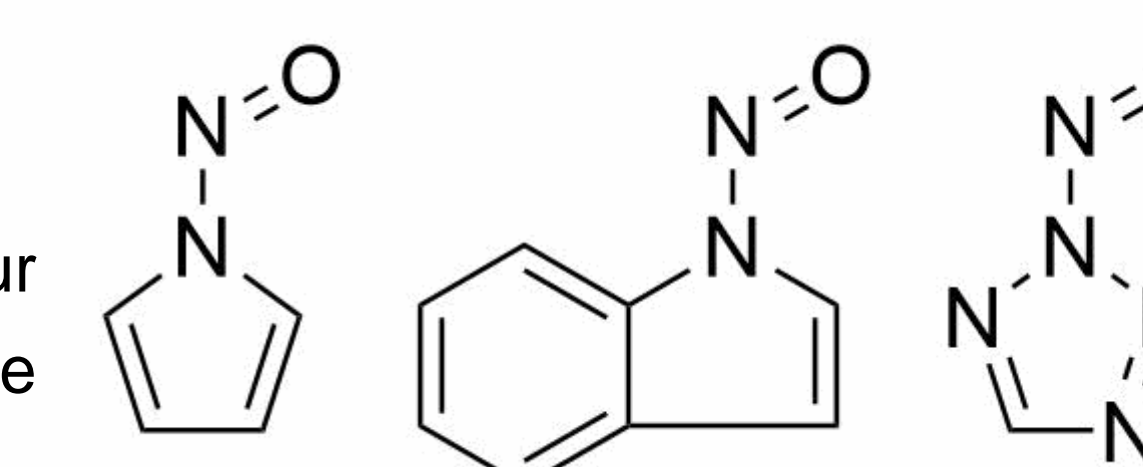
### 4. Nitrosamines with $\alpha$ -hydrogen & significant steric hindrance

- Mechanistically capable of undergoing  $\alpha$ -hydroxylation; however, metabolism it is not facile & these are observed to be non-carcinogenic.



### 5. Aromatic *N*-nitroso compound

- Unable to form diazonium ion, but mutagenicity may occur *via* *N*-nitroso reduction to a hydroxylamine, hence these have lower carcinogenic potential akin to aromatic amines.



### 6. Nitrosoureas & analogues

- Direct-acting carcinogens which have potency that conservatively warrants a position in the cohort of concern; however, because no metabolic activation is required, a negative result in the Ames test should be accepted.

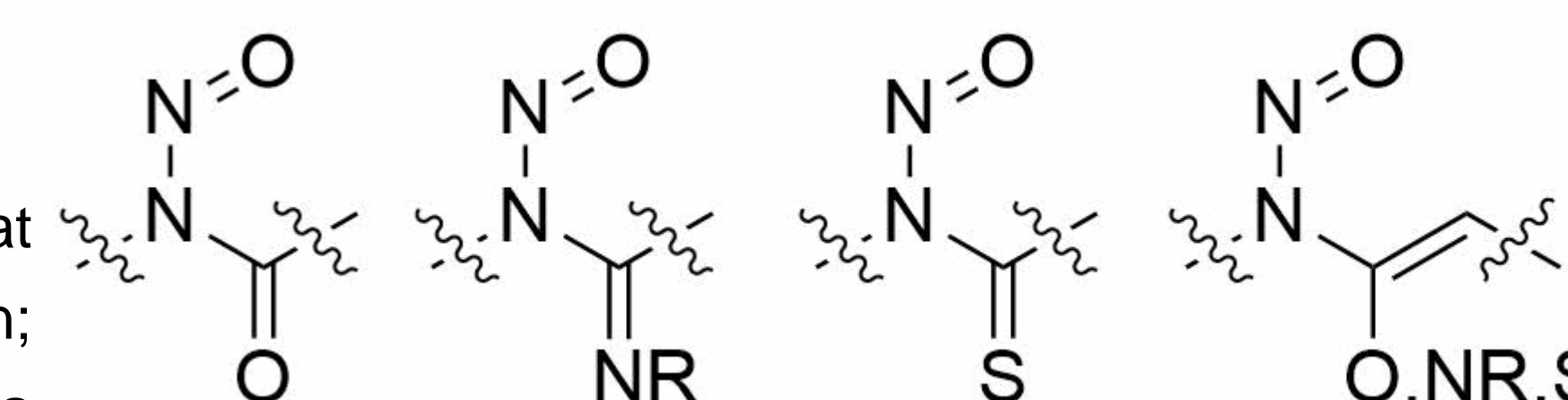


Fig. 3. Summary of carcinogenic potential of subclasses of *N*-Nitroso compounds.

[1] Thresher *et al.*, *Regul. Toxicol. Pharmacol.*, **2020**, 116, 104749. [2] Cross & Ponting, *Comput. Toxicol.*, **2021**, 100186.

[3] Thomas *et al.*, *Chem. Res. Toxicol.*, **2022**, 35, 1997-2013. [4] Ponting & Foster, *Org. Process Res. Dev.*, **2023**, Article ASAP.